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1) This term means two atoms with the same number of protons, but different numbers of neutrons. isotope
2) The element that has 12 protons (use your periodic table.) Mg
3) An atom that has lost or gained one or more electrons. ion
4) What is the charge on a proton? +15) This is the ability to move a mass (to do work). energy
6) About how many naturally-occurring (not man-made) elem...

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 $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
This can be interpreted as two molecules of hydrogen and one molecule of oxygen combining to form two water molecules. But how often do chemists limit their reactions to one or two molecules? Usually a reaction is done with an unimaginable number of molecules. How then do chemists k...

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